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Influence of doping elements (Cu and Fe) on the crystal structure and electrical resistivity of YNi_3 and $Y_{0.95}Ni_2$

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Two solid solutions on the basis of the binary compounds YNi₃ and $Y_{0.95}Ni_2$ were investigated by X-ray and electron probe microanalysis for the influence of Fe and Cu-doping on the crystal structure and electrical properties. Substitution of Cu for Ni in YNi₃ does not change the structure type (PuNi₃). The replacement of Ni by Cu in $Y_{0.95}Ni_{2-x}Cu_x$ up to x < 0.15 and the replacement of Ni by Fe in $Y_{0.95}Ni_{2-x}Fe_x$ up to x < 0.1 preserve the structure of the binary parent compound $Y_{0.95}Ni_2$ (TmNi₂ structure type, space group F-43m). Substitution by larger quantities of Cu and Fe destabilizes the TmNi₂ superstructure, which converts into the MgCu₂ structure with disordered Y vacancies. All the investigated alloys showed metallic-like conductivity following the Nordheim relation.

Yttrium intermetallic compounds / Doping / Crystal structure / Electrical resistivity

1. Introduction

Intermetallic compounds RM_n (R = rare earth, M = 3d-element, n = 2 or 3) exhibit a large variety of interesting physical properties. Some of these compounds also absorb large amounts of hydrogen, which induces remarkable changes of their physical properties. Intermetallics and their hydrides attract interest as objects for fundamental investigations and for practical application in the production of magnetic materials, hydrogen accumulators, metal hydride electrodes, *etc.* [1-3].

During the past years we have studied a number of binary RM_2 and RM_3 compounds doped with a third component (R' or M') with respect to the influence of the doping element on the crystal structure, electrical and magnetic properties and hydrogenation ability of these alloys. The results have been presented in [4-15].

Early reports on binary YNi_2 indicated that it belongs to the cubic $MgCu_2$ Laves phase structure [16]. It was recently shown that YNi_2 crystallizes in a superstructure of the cubic $MgCu_2$ Laves phase structure $(TmNi_2$ structure type) with the nominal composition $Y_{0.95}Ni_2$. This superstructure with space group F-43m is characterized by ordered Y vacancies in the 4a sites and a doubling of the lattice parameter a in comparison to the basic $MgCu_2$ structure [17].

Investigations of the electrical resistivity of YNi₂ [18] and of other transport properties (thermal conductivity, thermopower) of the RNi₂ series [1] evidenced anomalies for some of these compounds at high temperatures. Gratz et al. [19] later attributed high-temperature transport anomalies to structural phase transitions from the superstructure type (space group F-43m) at lower temperatures to the MgCu₂ structure type (space group Fd-3m), stable at higher temperatures. Our previous [5-9,11-13,15] were focused on the influence of substitution of Fe for Ni, or non-magnetic Y for the magnetic rare earths Er and Gd, on the crystal structure. electrical and magnetic properties, hydrogenation ability and kinetics of hydrogenation-dehydrogenation process of $R_{0.85}Y_{0.15}Ni_2$ and $RNi_{1.85}Fe_{0.15}$ (R = Gd, Er) alloys based on the binary RNi_2 (R = Gd, Er) compounds. The results of the crystal structure investigations showed that the parent compound RNi₂ and R/Y substituted $R_{0.85}Y_{0.15}Ni_2$ (R = Gd and Er) alloys belong to the TmNi₂ superstructure type, but the Fecontaining alloys GdNi_{1.85}Fe_{0.15} and ErNi_{1.85}Fe_{0.15} crystallize in the MgCu₂-type structure. All the investigated alloys showed metallic-like conductivity and exhibited long-range magnetic order [20]. Transition into a ferromagnetically ordered ground state was also apparent from the resistivity data. All the substituted alloys exhibited anomalies in their

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physical properties near the ferromagnetic phase transition temperatures. Increasing order of the magnetic structure with decreasing temperature was revealed by a decrease of the electrical resistivity.

Binary YNi₃ crystallizes in the PuNi₃-type rhombohedral structure (space group R-3m) [21]. The crystal structure of the RNi3 compounds can be presented as a stacking of RNi₅ (CaCu₅ Haucke phase) and RNi₂ (MgZn₂ Laves phase) structure fragments Hydrogen-absorbing RNi_3 intermetallic [22]. compounds (R = Y and rare earths) have already been known for more than three decades. However, increased interest in these compounds as hydrogen storage materials has appeared in recent years, mainly because of the prospect of electrochemical applications [23]. Among the RNi₃ compounds, YNi₃ displays the highest hydrogen sorption capacity. A partial replacement of Ni by Cr. Mn, Fe, Co, V [24], or Cu [10] decreases its hydrogen sorption capacity.

Single crystals of weakly itinerant ferromagnetic ($T_{\rm C} = 35 \, \text{K}$) YNi₃ [25] display a non-Fermi liquid (NLF) temperature dependence of the resistivity. The resistivity does not follow the T^2 -law [26], as would be expected from the conventional Fermi liquid theory.

Our previous systematic investigation of the phase equilibria in the Y–Cu–Ni system at 600° C [27,28] has shown the existence of limited solid solutions $Y_{1-\delta}Ni_{2-x}Cu_x$ (0 < δ < 0.05; 0 < x < 0.30) and $YNi_{3-x}Cu_x$ (x = 0.2 and x = 0.4) based on the binary compounds $Y_{1-\delta}Ni_2$ (0 ≤ δ ≤ 0.05) and YNi_3 . The present work focuses on investigations of the influence of Cu and Fe as doping elements on the structure and electrical resistivity of the binary compounds $Y_{0.95}Ni_2$ and YNi_3 .

2. Experimental details

Alloys were prepared by arc melting of starting elements with a purity not less than 99.9 wt.% under argon. The weight losses were less than 1 % of the total mass of the ingots. The alloys were homogenized in evacuated quartz ampoules at 870 K for 720 h. The samples were examined by X-ray powder diffraction (XRD) using a STOE STADI P diffractometer with Cu K_{α} -radiation. All the crystal structure calculations were performed by the Rietveld method with the programs FullProf [29] and CSD [30], using the compositions obtained electron by probe microanalysis. Qualitative and quantitative composition analyses on the bulk samples were performed with a scanning electron microscope REMMA-102-2. The temperature dependence of the electrical resistivity was studied by the four probe a.c.-bridge method in the temperature range 4-300 K.

3. Results and discussion

The XRD patterns of binary YNi₃ and the pseudobinary solid solution YNi_{3-x}Cu_x ($0 \le x \le 0.8$) were indexed in the PuNi₃-type structure (space group *R*-3*m*) and are presented in Fig. 1. The data show that the replacement of Ni by Cu in YNi₃ does not change the crystal structure of the solid solution. The lattice parameters increase with increasing Cu content (Table 1). The refined final atomic parameters for one alloy of the solid solution, YNi_{2.2}Cu_{0.8}, are presented in Table 2.

XRD patterns of the parent $Y_{0.95}Ni_2$ alloy and alloys of the solid solutions $Y_{0.95}Ni_{2-x}Cu_x$ ($0 \le x \le 0.3$)

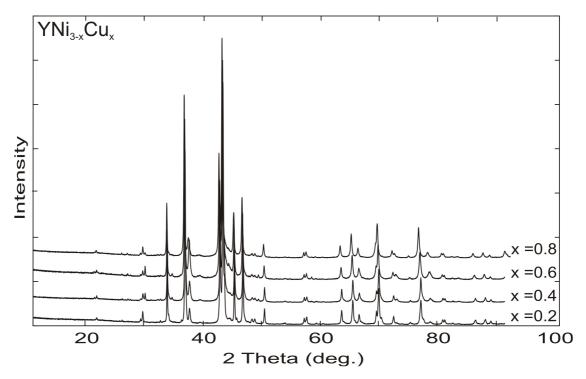


Fig. 1 X-ray powder diffraction patterns of alloys of the solid solution YNi_{3-x}Cu_x (PuNi₃-type structure).

Table 1 Lattice parameters of $YNi_{3-x}Cu_x$ ($0 \le x \le 0.8$; $PuNi_{3}$ -type structure, space group R-3m).

Composition	a (nm)	c (nm)	$V (\text{nm}^3)$
YNi ₃	0.4978(1)	2.4450(1)	0.5247(1)
$YNi_{2.8}Cu_{0.2}$	0.49837(1)	2.4454(1)	0.5258(1)
$YNi_{2.6}Cu_{0.4}$	0.49935(2)	2.4477(1)	0.5285(2)
$YNi_{2.4}Cu_{0.6}$	0.49966(1)	2.4486(1)	0.5293(1)
$YNi_{2.2}Cu_{0.8}$	0.50095(1)	2.4503(1)	0.5324(1)

Table 2 Refined atomic parameters, site occupations (G) and isotropic temperature coefficients (B_{iso}) for YNi_{2.2}Cu_{0.8} (PuNi₃-type structure, $R_B = 0.0587$).

Atom	Site	x	у	Z	G	$B_{\rm iso} \times 10^2 ({\rm nm}^2)$
Y1	3 <i>a</i>	0	0	0	1.0	1.31(5)
Y2	6 <i>c</i>	0	0	0.14041(9)	1.0	1.07(7)
M1	3b	0	0	1/2	0.753Ni+0.247Cu	1.51(7)
M2	6 <i>c</i>	0	0	0.33423(18)	0.750Ni+0.250Cu	1.53(11)
M3	18 <i>h</i>	0.8292(4)	0.1708(4)	0.58390(2)	0.736Ni+0.264Cu	1.32(6)

Table 3 Crystallographic parameters of solid solutions $Y_{0.95}Ni_{2-x}Cu_x$ ($0 \le x \le 0.3$) and $Y_{0.95}Ni_{2-x}Fe_x$ ($0 \le x \le 0.1$).

Composition	Structure type	Space group	a (nm)	$V(\text{nm}^3)$
Y _{0.95} Ni ₂ ^a	$TmNi_2$	F-43m	1.4347(2) ^a	2.9529(4) ^a
			1.4357 ^b	2.959 ^b
$Y_{0.95}Ni_{1.97}Cu_{0.03}$	$TmNi_2$	F-43m	1.43501(1)	2.95486(1)
$Y_{0.95}Ni_{1.92}Cu_{0.08}$	$TmNi_2$	F-43m	1.43541(1)	2.95881(1)
$Y_{0.95}Ni_{1.85}Cu_{0.15}$	$MgCu_2$	Fd-3m	0.71723(1)	0.36896(3)
$Y_{0.95}Ni_{1.7}Cu_{0.3}$	$MgCu_2$	Fd-3m	0.71823(1)	0.37031(2)
$YNi_{1.9}Cu_{0.2}^{b}$	$TmNi_2$	F-43m	1.4368 ^b	2.966 ^b
$YNi_{1.8}Cu_{0.4}^{b}$	$TmNi_2$	F-43m	1.4369 ^b	2.967 ^b
$Y_{0.95}Ni_{1.97}Fe_{0.03}$	$TmNi_2$	F-43m	1.43614(1)	2.96202(7)
$Y_{0.95}Ni_{1.95}Fe_{0.05}$	$TmNi_2$	F-43m	1.43639(3)	2.9636(2)
$Y_{0.95}Ni_{1.9}Fe_{0.1}$	$MgCu_2$	Fd-3m	0.71872(1)	0.37126(2)

^a present data; ^b data from [31]

and $Y_{0.95}Ni_{2-x}Fe_x$ ($0 \le x \le 0.1$) are presented in Figs. 2 and 3. A detailed analysis of the crystal structure of these solid solutions showed that the replacement of Ni by Cu in $Y_{0.95}Ni_{2-x}Cu_x$ up to x < 0.15 (< 5 at.% Cu) and the substitution of Ni by Fe in Y_{0.95}Ni_{2-x}Fe_x up to x < 0.1 (< 3 at.% Fe) preserve the structure of the binary parent compound Y_{0.95}Ni₂ (TmNi₂ structure type, space group F-43m). The presence of superstructure reflections (see Figs. 2 and 3) indicates that all the samples of the solid solution belong to the cubic structure type with doubled cell parameter 2a (where a is the unit cell parameter of the MgCu₂-type structure). The lattice parameters increase with increasing Cu and Fe content (see Table 3). Higher contents of Cu and Fe lead to a phase transition from the TmNi₂ superstructure to the MgCu₂ structure type. There are no visible superstructure lines in the XRD patterns of $Y_{0.95}Ni_{1.7}Cu_{0.3}$ and $Y_{0.95}Ni_{1.9}Fe_{0.1}$ (see Figs. 2 and 3). However, according to Paul-Boncour et al. [31] the pseudo-binary solid solution Y(Ni,Cu)₂ adopts the superstructure within the homogeneity range up to about 20 at.% Cu (x=0.6). The difference between the present data and those of [31] may be attributed to the different annealing temperatures applied in the two investigations, 870 K and 1023 K, respectively. Refined lattice parameters of the solid solution series $Y(Ni,Cu)_2$ are summarized in Table 3. The atomic parameters of $Y_{0.95}Ni_{1.92}Cu_{0.08}$ (TmNi2-type structure) and $Y_{0.95}Ni_{1.7}Cu_{0.3}$ (MgCu₂-type structure) alloys are presented in Tables 4 and 5.

Electrical resistivity measurements were carried out to supplement the crystallographic studies by the information revealed from the scattering of conduction electrons in the investigated solid solutions In solid solutions of fully miscible isostructural nonmagnetic elements, *e.g.* alloys of the noble metals Ag and Au, the low-temperature resistivities are dominated by disorder scattering of the conduction electrons. Accordingly, the residual resistivities of the solid solution $Au_{1-x}Ag_x$ follow the simple Nordheim relation, $\rho_0(x) = C \cdot x(1-x)$ [32]. Deviations of $\rho_0(x)$ from the Nordheim rule are expected in the presence

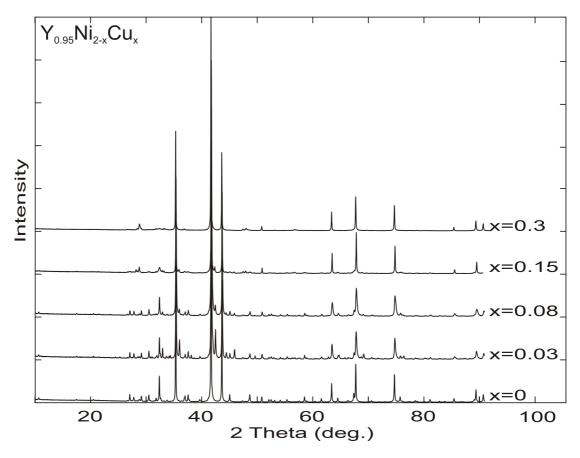


Fig. 2 X-ray powder diffraction patterns of alloys of the solid solution $Y_{0.95}Ni_{2-x}Cu_x$.

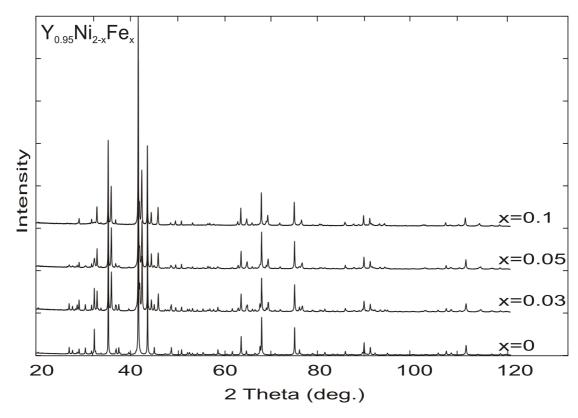


Fig. 3 X-ray powder diffraction patterns of alloys of the solid solution Y_{0.95}Ni_{2-x}Fe_x.

Table 4 Refined atomic parameters,	site occupation	(G) and isotropic	temperature	coefficients	$(B_{\rm iso})$	for
$Y_{0.95}Ni_{1.92}Cu_{0.08}$ (TmNi ₂ -type structure,	$R_{\rm B} = 0.052$).					

Atom	Site	х	у	z	G	$B_{\rm iso} \times 10^2 ({\rm nm}^2)$
Y1	4 <i>a</i>	0	0	0	0.401(8)	1.49(1)
Y2	4b	1/2	1/2	1/2	1.0	0.65(1)
Y3	16 <i>e</i>	0.10308(5)	x	x	1.0	0.69(1)
Y4	16 <i>e</i>	0.62606(7)	X	X	1.0	0.52(1)
Y5	24 <i>g</i>	0.0078(1)	1/4	1/4	1.0	0.93(1)
M1	16 <i>e</i>	0.3118(2)	x	x	0.96 Ni + 0.04 Cu	0.52(1)
M2	16 <i>e</i>	0.8118(2)	x	x	0.96 Ni + 0.04 Cu	0.99(1)
M3	48h	0.0654(1)	x	0.8089(2)	0.96 Ni + 0.04 Cu	0.57(1)
M4	48h	0.0619(1)	X	0.3118(2)	0.96 Ni + 0.04 Cu	0.61(1)

Table 5 Refined atomic parameters, site occupation (*G*) and isotropic temperature coefficients (B_{iso}) for $Y_{0.95}Ni_{1.7}Cu_{0.3}$ (MgCu₂-type structure, $R_B = 0.032$).

Atom	Site	x	у	z	G	$B_{\rm iso} \times 10^2 (\rm nm^2)$
Y	8b	3/8	3/8	3/8	0.95	1.06(1)
M	16 <i>c</i>	0	0	0	0.85 Ni + 0.15 Cu	1.04(1)

of magnetic elements and, in particular, for solid solutions undergoing changes of their crystal structures and/or magnetic ground states.

Temperature dependent electrical resistivities, $\rho(T)$, of polycrystalline samples of the binary parent compound YNi₃ and the solid solutions YNi_{3-r}Cu_r (x = 0.2, 0.4, 0.6, and 0.8) are displayed in Fig. 4a. The resistivities $\rho(T)$ of all the samples in this series exhibit a rather simple, metallic-like temperature dependence. Only in the case of weakly ferromagnetic YNi₃ the resistivity data exhibit a sharp kink at about 32 K, which indicates a phase transition to a ferromagnetic ground state (previously reported by Gignoux et al. [25]). The present results obtained for polycrystalline YNi3 are in reasonable agreement with the earlier single-crystal studies [25]. The substitution of Cu for Ni is expected to weaken and eventually suppress long-range ferromagnetic order. The evolution of the composition dependence of the residual resistivity $\rho_0(x)$, *i.e.* the resitivity values $\rho(4.2 \text{ K})$ of each composition YNi_{3-x}Cu_x, collected in Fig. 4b, reveals a marked increase of ρ_0 from YNi₃ to YNi_{2.8}Cu_{0.2} by about 35 $\mu\Omega$ cm, but a significantly smaller variation of $\rho_0(x)$ within the solid solution YNi_{3-r}Cu_r, which approximately follows a modified Nordheim rule, $\rho_0(x) = A + C \cdot x(3-x)$, where A is a constant off-set and C the factor for disorder scattering in the (Ni,Cu) sublattice. The dashed line in Fig. 4b shows a fit using the Nordheim rule, which yields $A \approx 40 \,\mu\Omega$ cm and $C \approx 20 \,\mu\Omega$ cm. The 40 $\mu\Omega$ cm offset of the residual resistivities of the Cu doped samples most likely originates from incoherent magnetic scattering of the conduction electrons, i.e. some kind of spin disorder scattering induced by substitutional disorder, which suppresses the longrange ferromagnetic order of the parent compound YNi₃.

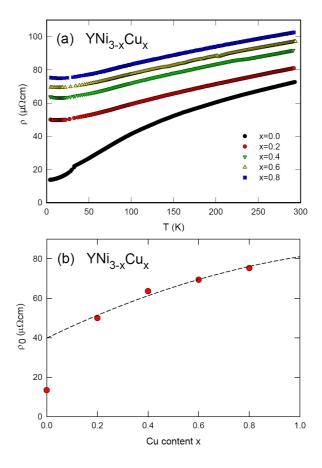


Fig. 4 Temperature dependent electrical resistivity $\rho(T)$ (a) and composition dependent residual resistivity $\rho_0(x)$ (b) of solid solutions $YNi_{3-x}Cu_x$; the dashed line shows a fit according to the Nordheim rule.

In case of the solid solution $Y_{0.95}Ni_{2-x}Cu_x$, all the compositions, including the binary parent compound $Y_{0.95}Ni_2$, show Pauli paramagnetic ground state and the electrical resistivities displayed in Fig. 5a exhibit a simple metallic temperature dependence, which is dominated by the scattering of conduction electrons on static lattice defects and phonons and consequently follows the simple Bloch-Grüneisen (BG) law

$$\rho(T) = \rho_0 + B \frac{T^5}{\Theta_D^6} \int_0^{\theta_D/T} \frac{x^5}{(e^x - 1)(1 - e^{-x})} dx,$$

yielding the values of residual resistivity shown in Fig. 5b, moderate variations of the Debye temperature $\Theta_{\rm D} \approx 220 \pm 30 \, \rm K$ and an electron-phonon coupling factor B in the range 6-10 m Ω cm K. The moderate reduction of the temperature dependence of the resitivity of Y_{0.95}Ni_{1.7}Cu_{0.3} is attributed to the larger absolute resistivity exceeding 150 $\mu\Omega$ cm, which leads to deviations from the Matthiessen rule and consequently to a flattening of $\rho(T)$ (see e.g. [33]). The residual resistivity, $\rho_0(x)$, of $Y_{0.95}Ni_{2-x}Cu_x$ in Fig. 5b reveals a significant discontinuity within the series, *i.e.* a marked increase of ρ_0 between $Y_{0.95}Ni_{1.92}Cu_{0.08}$ and $Y_{0.95}Ni_{1.85}Cu_{0.15}$, which coincides with the vanishing of the superstructure reflections in Fig. 2. i.e. with the change of the crystal structure from the TmNi₂ type to the MgCu₂ type. The different trends of $\rho_0(x)$ in the TmNi₂-type and MgCu₂-type ranges of the solid solution Y_{0.95}Ni_{2-x}Cu_x are emphasized by dashed and dotted lines [Nordheim fits, $\rho_0(x) = A + C \cdot x(2-x)$], respectively. According to these fits, the loss of the superstructure ordering of Y vacancies is accompanied by an increase of the A parameter from $10 \,\mu\Omega$ cm to $22 \,\mu\Omega$ cm and an increase of the factor C from $140 \,\mu\Omega$ cm to 225 $\mu\Omega$ cm. Magnetic effects do not seem to play any role in the solid solution Y_{0.95}Ni_{2-x}Cu_x, because all the samples remain paramagnetic.

The most significant changes of the electrical resistivity are observed in the solid solution $Y_{0.95}Ni_{2-x}Fe_x$ (see Fig. 6). Only the binary parent compound Y_{0.95}Ni₂ exhibits a simple BG-like metallic behavior (as mentioned above), whereas the Fe-doped solid solutions initially display an one order of magnitude stronger increase of the residual resistivity ρ_0 as compared to $Y_{0.95}(Ni,Cu)_2$ and $Y(Ni,Cu)_3$, and also a marked increase of the slope of $\rho(T)$ at the lowest temperatures with an approximate T-linear temperature dependence. At higher Fe contents the residual resistivity drops and for Y_{0.95}Ni_{1.7}Fe_{0.3} the low-temperature behavior has changed back to the more common T-square dependence. The Fe-richest composition $Y_{0.95}Ni_{1.5}Fe_{0.5}$ exhibits a kink in $\rho(T)$ at about 80 K. All these marked changes refer to magnetic effects caused by the Fe doping, which may initially lead to a magnetic spin or cluster glass ground state (for $Y_{0.95}Ni_{2-x}Fe_x$, x = 0.05 and 0.1), and eventually to long-range ferromagnetic order, which is in line with the T^2 dependence of $Y_{0.95}Ni_{1.7}Fe_{0.3}$ and Y_{0.95}Ni_{1.5}Fe_{0.5} and explains, in particular, the kink in $\rho(T)$ observed for the latter. More detailed magnetic studies are in progress to clarify the magnetic ground states in the solid solution series $Y_{0.95}Ni_{2-x}Fe_x$.

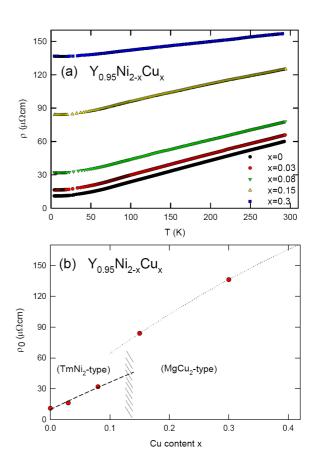


Fig. 5 Temperature dependent electrical resistivity $\rho(T)$ (a) and composition dependent residual resistivity $\rho_0(x)$ (b) of solid solutions $Y_{0.95}Ni_{2-x}Cu_x$; the dashed and dotted lines indicate fits according to the Nordheim rule for the TmNi₂-type and MgCu₂-type solid solutions, respectively.

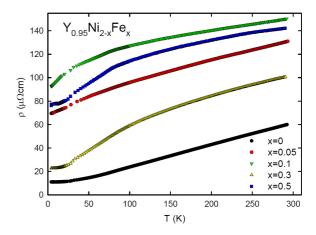


Fig. 6 Temperature dependent electrical resistivity of solid solutions $Y_{0.95}Ni_{2-x}Fe_x$.

4. Conclusions

Substitution of Cu for Ni in the binary compound YNi₃ (PuNi₃-type structure, space group R-3m) results in the existence of a limited solid solution YNi_{3-x}Cu $_x$ ($0 \le x \le 0.8$). The lattice parameters increase with increasing Cu content. The replacement of Ni by small quantities of Cu in Y_{0.95}Ni_{2-x}Cu $_x$ up to x < 0.15 and substitution of Fe for Ni in Y_{0.95}Ni_{2-x}Fe $_x$ up to x < 0.1 preserve the structure of the binary parent compound Y_{0.95}Ni₂ (TmNi₂ structure type). The replacement of Ni by larger quantities of Cu and Fe destabilizes the TmNi₂ superstructure, which converts into the MgCu₂ structure with a constant amount of Y vacancies.

results of the electrical resistivity measurements of the Cu-doped solid solutions are mostly in line with the usual substitutional disorder effect as suggested by the Nordheim rule. Deviations from the simple Nordheim-type variation of the residual resistivity are clearly connected to the loss of weak ferromagnetic order in the solid solution YNi_{3-x}Cu_x and to the loss of superstructure ordering in Y_{0.95}Ni_{2-x}Cu_x. In the case of the solid solution $Y_{0.95}Ni_{2-x}Fe_x$, large changes of the temperature dependent electrical resistivity are observed and refer to effects of Fe magnetism, which for lower Fe concentrations is most likely of the spin glass type and long-range ferromagnetic for larger Fe contents.

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