# The first quaternary intermetallide with Sc<sub>3</sub>Ce<sub>2</sub>Si<sub>4</sub>-type structure

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The crystal structure of Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67</sub>Si<sub>1.33</sub> was solved and refined from X-ray single-crystal diffraction data. It is isotypic to Sc<sub>3</sub>Ce<sub>2</sub>Si<sub>4</sub>: Pearson symbol *oP*36, space group *Pnma*, a = 6.848(2), b = 13.299(5), c = 7.032(3) Å, Z = 4; R = 0.0329, wR = 0.0817 (for 608 independent reflections with  $F > 2\sigma(F)$  and 50 variables). The structure of the new quaternary compound is partly disordered and characterized by statistic distribution of the smaller atoms, Ge and Si, which form pairs with the shortest distances 2.677(3) and 2.561(4) Å. The structure of Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67</sub>Si<sub>1.33</sub> can be described as a stacking of slabs along the crystallographic direction [010]. The slabs are formed by Ta-centered rhombic dodecahedra and are connected *via* common faces.

Germanium / Silicon / Tantalum / Thulium / X-ray single crystal diffraction / Crystal structure

### Introduction

46 representatives of the orthorhombic structure type Sc<sub>3</sub>Ce<sub>2</sub>Si<sub>4</sub> (Pearson symbol oP36, space group *Pnma* [1]) are known so far, according to Pearson's Crystal Data [2]. This structure is formed in ternary systems with silicon: Sc-{La-Nd, Sm, Gd-Tm}-Si; Y-Ce-Si, Nb-{Sc, U}-Si; Mo-{Sc, Lu, Hf}-Si, W-Sc-Si, and germanium: Ca-{La, Ce}-Ge; Sc-{La-Nd, Sm, Gd}-Ge; Mo-Sc-Ge, Ti-{Gd-Tm, V–Sc–Ge; Y–Ce–Ge; Zr-Ta-Ge; Lu}–Ge; Nb-{Y, Gd-Lu}-Ge (the order of the chemical elements in the systems corresponds to the order of the chemical elements in the formula of the structure type Sc<sub>3</sub>Ce<sub>2</sub>Si<sub>4</sub>). Almost half of these compounds, 22 compounds, are Sc-containing compounds. In the structures of the compounds with rare-earth metals, the Sc atoms occupy the sites of Sc in the prototype, whereas in the compounds with other *d*-elements (Nb, Mo, W, V) the Sc atoms occupy the site of Ce in the prototype. In [3] the crystal structures of the compounds  $Nb_3R_2Ge_4$  (R = Y, Gd-Er, Lu) were assigned to the closelv related structure type Zr2Nb3Ge4 ([NbZr<sub>2</sub>]Nb<sub>2</sub>Ge<sub>4</sub>) [4]. The ternary types Sc<sub>3</sub>Ce<sub>2</sub>Si<sub>4</sub> and Zr<sub>2</sub>Nb<sub>3</sub>Ge<sub>4</sub> are ordered substitution derivatives of the binary type  $Gd_5Si_4$  [5], which is a branch of the type Sm<sub>5</sub>Ge<sub>4</sub> [6].

In this work we present the results of a structural investigation of the first representative of the  $Sc_3Ce_2Si_4$ -type found in a quaternary system, Ta-Tm-Si-Ge.

# Experimental

A sample of nominal composition  $Tm_{33.3}Ge_{61.7}Si_5$  was synthesized from high-purity elements ( $Tm \ge 99.83$ ,  $Ge \ge 99.999$ , and  $Si \ge 99.999$  wt.%) by arc melting in a water-cooled copper crucible with a tungsten electrode under a purified argon atmosphere (using Ti as a getter). The weight loss during the preparation of the alloy was less than 0.5 % of the total mass, which was 1 g. The sample was annealed at 600°C in a quartz ampoule under vacuum for one month and subsequently quenched in cold water. Single crystals were grown in a high-frequency furnace. The alloy, placed in a tantalum tube, was heated to a light orange color (until a flash was observed), kept for 30 minutes at this temperature, and then slowly (1 K/min) cooled.

Search and selection of a single crystal were carried out using a Leica 420i electron microscope. For further research, the single crystal was glued onto a glass capillary and tested for 5 min on a singlecrystal diffractometer Oxford Diffraction Xcalibur 3, Sapphire 3 CCD Detector (Mo Ka radiation). The obtained diffraction pattern was analyzed using the CrysAlis computer program package [7], which allows evaluating the quality of the single crystal and the parameters of the unit cell. Possible space groups were determined from an analysis of the systematic extinctions. The crystal structure was solved by direct methods and subsequent least-squares refinements were carried out with the SHELX-97 [8] program package. The chemical composition of the single crystal was checked by means of energy-dispersive X-ray analysis, using a scanning electron microscope JEOL SEM 5900LV. The structure drawings were made with the program ATOMS [9].

# **Results and discussion**

A preliminary analysis of the diffraction data showed that the investigated crystal belongs to the orthorhombic system. The final refinement was performed in the anisotropic approximation in space group Pnma and led to the composition Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67(5)</sub>Si<sub>1.33(5)</sub>, which agrees well with the results of the EDX analysis of the crystal (see Table 1). Details of the data collection and final structure refinement are given in Table 2. The new quaternary compound, Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67</sub>Si<sub>1.33</sub>, crystallizes with a Sc<sub>3</sub>Ce<sub>2</sub>Si<sub>4</sub>-type structure: the Tm atoms occupy the positions of the Ce atoms (Wyckoff position 8d of the space group Pnma), the Ta atoms occupy the positions of the Sc atoms (8d and 4c), whereas the Ge and Si atoms statistically occupy the positions of the Si atoms (8d, 4c, and 4c). Two of the three sites occupied by the statistical mixture Ge/Si are mainly occupied by Ge atoms, more than 70 %. The refined coordinates, atomic site occupancies, and

displacement parameters for Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67</sub>Si<sub>1.33</sub> are listed in Table 3 and Table 4. The positional parameters were standardized with the program STRUCTURE TIDY [10]. The interatomic distances within the coordination polyhedra are listed in Table 5. The shortest distances are observed between Ge/Si atoms of the sites *M*1, *M*2 and *M*3:  $\delta_{M1-M1} =$ 2.677(3),  $\delta_{M2-M3} = 2.561(4)$  Å. A projection of the structure Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67</sub>Si<sub>1.33</sub> along the crystallographic direction [100] is presented in Fig. 1.

In the structure of Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67</sub>Si<sub>1.33</sub>, the Tm atoms have the largest coordination polyhedra, which can be described as a pentagonal prism consisting of one Tm, four Ta, and five Ge/Si atoms, and eight additional atoms of different kinds. The Ta atoms are situated in slightly smaller polyhedra: 16-vertex Frank-Kasper polyhedra (site Ta1) and rhombic dodecahedra (Ta2). In the case of site Ta1, the coordination polyhedron consists of five Tm, five Ta and six Ge/Si atoms, whereas the composition of the dodecahedron is  $Tm_4Ta_4M_6$ . The Ge and Si atoms center trigonal prisms of compositions Tm<sub>3</sub>Ta<sub>3</sub> (M1),  $Tm_2Ta_4$  (M2) and  $Tm_4Ta_2$  (M3). All of the prisms are three-capped; in the case of sites M2 and M3 by one Ge/Si and two Ta atoms, in the case of site M1 by atoms of the three/four elements (Tm, Ta, Ge/Si).

Table 1 Results of the EDX analysis of the single crystal used in the experiment.

Element	Content (at.%)	Accuracy of the analysis (at.%)
Tm	21.87	$\pm 2.7$
Ta	31.65	$\pm 2.1$
Ge	30.59	$\pm 3.6$
Si	15.89	$\pm 3.9$

Composition from refinement	$1 m_2 1 a_3 Ge_{2.67} Si_{1.33}$
Molar mass $M$ (g mol <sup>-1</sup> )	1111.66
Space group	Pnma
Cell parameters $a, b, c$ (Å), $V$ (Å <sup>3</sup> )	6.848(2), 13.299(5), 7.032(3), 640.4(4)
Formula units per unit cell Z	4
Density $D_x$ (Mg m <sup>-3</sup> )	11.530
Absorption coefficient $\mu$ (mm <sup>-1</sup> )	90.866
Radiation, wavelength (Å)	Μο Κα, 0.71073
Number of reflections for cell parameters	608
Temperature for data collection (K)	293(2)
Crystal color	gray, metallic
Collection mode	$\omega$ oscillation scans
Number of measured / independent reflections	18679 / 767
Number of reflections with $F > 2\sigma(F)$	608
R <sub>int</sub>	0.0736
Range of $h, k, l$	$-8 \le h \le 8, -17 \le k \le 17, -9 \le l \le 9$
Refinement on	$F^2$
Reliability factors R, wR (for $F \ge 2\sigma(F)$ )	0.0407 (0.0329), 0.0840 (0.0817)
Goodness of fit S	1.126
Number of refined parameters	50
Weighting scheme	$w = 1/[(\sigma F_o)^2 + (0.0527P)^2]$ , where $P = (F_o^2 + 2F_c^2)/3$
Extinction coefficient	0.00045(8)
Residual electron density $\Delta \rho_{\text{max}} / \Delta \rho_{\text{min}}$ (e Å <sup>-3</sup> )	3.472 / -3.393

Table 2 Crystal data and details of the data collection and final refinement for Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67</sub>Si<sub>1.33</sub>.

**Table 3** Atom coordinates and site occupancies for  $Tm_2Ta_3Ge_{2.67(5)}Si_{1.33(5)}$ : structure type  $Sc_3Ce_2Si_4$ , *oP*36, *Pnma*, *a* = 6.848(2), *b* = 13.299(5), *c* = 7.032(3) Å.

Site	Wyckoff position	x	у	Z	Occupancy
Tm	8 <i>d</i>	0.99647(8)	0.59509(5)	0.17285(9)	1
Tal	8 <i>d</i>	0.34034(8)	0.12638(4)	0.17317(8)	1
Ta2	4 <i>c</i>	0.16699(11)	1⁄4	0.48903(11)	1
Gel	8 <i>d</i>	0.1748(2)	0.04066(13)	0.4637(2)	0.771(12)
Sil	8 <i>d</i>	0.1748(2)	0.04066(13)	0.4637(2)	0.229(12)
Ge2	4 <i>c</i>	0.0411(5)	1⁄4	0.1199(5)	0.391(16)
Si2	4 <i>c</i>	0.0411(5)	1⁄4	0.1199(5)	0.609(16)
Ge3	4 <i>c</i>	0.3045(3)	1⁄4	0.8615(4)	0.733(15)
Si3	4 <i>c</i>	0.3045(3)	1⁄4	0.8615(4)	0.267(15)

Table 4 Equivalent and anisotropic displacement parameters (Å<sup>2</sup>) for Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67</sub>Si<sub>1.33</sub>.

Site	$U_{ m eq}$	$U_{11}$	$U_{22}$	$U_{33}$	$U_{12}$	$U_{13}$	$U_{23}$
Tm	0.0047(2)	0.0059(3)	0.0029(3)	0.0054(4)	0.0001(2)	-0.0001(2)	0.0003(2)
Tal	0.0042(2)	0.0060(3)	0.0011(3)	0.0054(3)	-0.0003(2)	0.0000(2)	0.0002(2)
Ta2	0.0038(2)	0.0044(4)	0.0019(4)	0.0050(4)	0.000	-0.0002(3)	0.000
Gel	0.0044(6)	0.0046(9)	0.0019(9)	0.0067(10)	-0.0017(7)	0.0014(6)	0.0001(6)
Si1	0.0044(6)	0.0046(9)	0.0019(9)	0.0067(10)	-0.0017(7)	0.0014(6)	0.0001(6)
Ge2	0.0051(11)	0.0088(18)	0.0022(18)	0.0042(18)	0.000	-0.0015(13)	0.000
Si2	0.0051(11)	0.0088(18)	0.0022(18)	0.0042(18)	0.000	-0.0015(13)	0.000
Ge3	0.0044(8)	0.0052(13)	0.0023(13)	0.0058(13)	0.000	-0.0008(10)	0.000
Si3	0.0044(8)	0.0052(13)	0.0023(13)	0.0058(13)	0.000	-0.0008(10)	0.000



**Fig. 1** Projection of the structure of  $Tm_2Ta_3Ge_{2.67}Si_{1.33}$  along [100] emphasizing the pairs of M (Ge/Si) atoms.

The structure of  $Tm_2Ta_3Ge_{2.67}Si_{1.33}$  can be represented as a stacking of layers of rhombic dodecahedra along the crystallographic direction [010]. Within the layers, the dodecahedra are connected *via* common faces (Fig. 2).

The structure type  $Sc_3Ce_2Si_4$  is a ternary substitution derivative of the binary structure types  $Sm_5Ge_4$  [6] and  $Gd_5Si_4$  [5], where two of the three sites occupied by rare-earth atoms in the binary types (Wyckoff positions 8*d* and 4*c* of space group *Pnma*) are occupied by Sc atoms and the last one (8*d*) by Ce atoms. The structure type  $Gd_5Si_4$  is closely related to the type  $Sm_5Ge_4$  and is sometimes considered as a branch of the latter [6]. In  $Gd_5Si_4$  and  $Sc_3Ce_2Si_4$  all of the Si atoms form pairs ( $\delta(Si1-Si1) = 2.48$  and 2.54 Å,  $\delta(Si2-Si3) = 2.48$  and 2.61 Å, respectively, site names standardized) [11]. In  $Sm_5Ge_4$  significantly longer distances occur within the Ge1-Ge1 pairs (3.71 Å) than within the Ge2-Ge3 pairs (2.66 Å).



Fig. 2 Stacking of rhombic dodecahedra in the structure of  $Tm_2Ta_3Ge_{2.67}Si_{1.33}$ .

In the case of the compound  $Tm_2Ta_3Ge_{2.67}Si_{1.33}$ , synthesized here, the *M*1-*M*1 distances (2.60 Å) are only slightly longer than the *M*2-*M*3 distances (2.56 Å), and both must be considered to present partly covalent character. A disordered arrangement of the Ge and Si atoms was observed, however, the Si atoms show a preference for site *M*2.

### Conclusions

The  $Tm_2Ta_3Ge_{2.67}Si_{1.33}$  compound is the first quaternary intermetallide found to crystallize with a  $Sc_3Ce_2Si_4$ -type structure. The structure is disordered in relation to the Ge and Si atoms. It belongs to a family of structures with trigonal prismatic coordination of

Atoms		$\delta$ (Å)			toms	$\delta$ (Å)	
Tm	- <i>M</i> 1	2.785(2)		Ta2	- M2	2.674(3)	
	- <i>M</i> 3	2.804(2)			- <i>M</i> 3	2.696(3)	NI NI
	- <i>M</i> 1	2.904(2)			- <i>M</i> 2	2.735(4)	
	- <i>M</i> 2	2.923(3)			- <i>M</i> 3	2.783(3)	
	- <i>M</i> 3	2.924(2)			- 2 <i>M</i> 1	2.790(2)	
	- <i>M</i> 1	2.989(2)	Tm		- 2Ta1	3.001(2)	
	- <i>M</i> 1	3.006(2)	Ta2 a1		- 2Ta1	3.007(2)	Tm
	- Ta2	3.339(2)	MI M3		- 2Tm	3.339(2)	
	- Ta2	3.351(2)	M3 Tal		- 2Tm	3.351(2)	
	- Ta1	3.315(2)		M1	- Tal	2.599(2)	
	- Ta1	3.378(2)			- Tal	2.668(2)	
	- Tm	3.508(3)	Ta2 Ta1		- <i>M</i> 1	2.677(3)	Tm
	- 2Tm	3.592(3)	•		- Tal	2.734(2)	
	- Ta1	3.711(2)			- Tm	2.785(2)	
	- Tal	3.715(2)			- Ta2	2.790(2)	
	- Tal	3.771(2)			- Tm	2.904(2)	Ta2
	- Tm	4.120(3)			- Tm	2.989(2)	
Ta1	- <i>M</i> 2	2.590(3)			- Tm	3.006(2)	
	- <i>M</i> 1	2.599(2)		M2	- <i>M</i> 3	2.561(4)	
	- <i>M</i> 2	2.654(3)			- 2Ta1	2.590(3)	Tm -M3
	- <i>M</i> 1	2.668(2)	Tm		- 2Ta1	2.654(3)	Ta2
	- <i>M</i> 1	2.734(2)			- Ta2	2.674(3)	
	- <i>M</i> 3	2.751(2)	Tm M3		- Ta2	2.735(4)	Tal
	- Ta2	3.001(2)			- 2Tm	2.924(3)	Ta2
	- Ta2	3.007(2)	Tm Tal Tal				
	- Tal	3.288(2)	Tal al M	М3	- <i>M</i> 2	2.561(4)	Ta2
	- Tm	3.315(2)	Ta2		- Ta2	2.696(3)	
	- Tm	3.378(2)	Tm		- 2Tal	2.751(2)	Tm
	- 2Tal	3.590(2)			- Ta2	2.783(3)	
	- Tm	3.711(2)			- 2Tm	2.804(2)	Ta2
	- Tm	3.715(2)			- 2Tm	2.924(2)	Tal
	- Tm	3.771(2)					•

Table 5 Interatomic distances and coordination polyhedra in Tm<sub>2</sub>Ta<sub>3</sub>Ge<sub>2.67</sub>Si<sub>1.33</sub>.

the smaller atoms (Ge and Si), whereas the coordination polyhedra of the Tm and Ta atoms (two sites for the latter) are larger and correspond to 18-, and 16-, and 14-vertex polyhedra. The structure can be viewed as a stacking of layers, formed by interconnected Ta-centered 14-vertex polyhedra (rhombic dodecahedra), along the crystallographic direction [010].

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#### References

- [1] I.R. Mokra, O.I. Bodak, E.I. Gladyshevskii, *Kristallografiya* 24 (1979) 1274-1276.
- [2] P. Villars, K. Cenzual (Eds.), Pearson's Crystal Data – Crystal Structure Database for Inorganic Compounds, ASM International, Materials Park, OH, USA, Release 2021/22.

- [3] B. Reker, S.F. Matar, U.C. Rodewald, R.D. Hoffmann, R. Pöttgen, Z. Naturforsch., B: J. Chem. Sci. 68 (2013) 625-634.
- [4] Y.D. Seropegin, V.V. Tabachenko, M.G. Myskiv, *Kristallografiya* 29 (1984) 161-163.
- [5] J.E. Iglesias, H. Steinfink, J. Less-Common Met. 26 (1972) 45-52.
- [6] G.S. Smith, Q.C. Johnson, A.G. Tharp, *Acta Crystallogr.* 22 (1967) 269-272.
- [7] Oxford Diffraction Xcalibur System, User Manual, CrysAlis Software Package, Oxfordshire, UK, 2001.
- [8] G.M. Sheldrick, Acta Crystallogr., Sect. C: Struct. Chem. 71 (2015) 3-8.
- [9] E. Dowty, ATOMS A Computer Program for Displaying Atomic Structures, Kingsport, TN, USA, 1999.
- [10] L.M. Gelato, E. Parthé, J. Appl. Crystallogr. 20 (1987) 139-143.
- [11] E. Parthé, L. Gelato, B. Chabot, M. Penzo, K. Cenzual, R. Gladyshevskii, TYPIX – Standardized Data and Crystal Chemical Characterization of Inorganic Structure Types, Gmelin Handbook of Inorganic and Organometallic Chemistry, Springer-Verlag, Berlin, 1993, Vol. 1, pp. 207-216.